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Novel sorbent shows promising financial results on P recovery from sludge water

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Credit Author Statement

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broken cellulose casings



pyrolysis



activation via  $\text{CaCl}_2$



sludge water



CaP - enriched biochar



soil application

Journal Pre-proof

1 Title

2 Novel sorbent shows promising financial results on P recovery from sludge water

3

4 Abstract

5 For several decades, researchers have been struggling to obtain minimum phosphorus (P) capture  
6 costs to meet the parameters for discharging wastewater into the watercourse. Findings from  
7 ongoing practices suggest that the Modified University of Cape Town process is currently the  
8 cheapest P capture method in the USA, whereas struvite precipitation seems to be the most cost  
9 effective method in the rest of the developed world. P sorption via biochars is becoming  
10 widespread in developing countries because this technique allows for the turning of voluminous  
11 biowaste into fertilizer with soil improving properties. Nevertheless, the reliability of this  
12 technology fluctuates throughout the year according to biowaste characteristics. For the first time,  
13 it has been proposed to use broken cellulose casings, which are readily available in increasing  
14 quantities worldwide. The sorbent obtained was subsequently activated by calcium chloride  
15 ( $\text{CaCl}_2$ ), whose cost is irrelevant as it would be used for agronomical purposes anyway. Pilot scale  
16 experiments show that this novel sorbent is capable of capturing  $31.8 \text{ kg P t}^{-1}$  from sludge water  
17 that contains  $52.5 \text{ mg}$  of extractable  $\text{P L}^{-1}$ . More importantly, it was reported that the novel sorbent  
18 captures P, mostly in calcium phosphates (CaP) forms ( $191.5 \text{ g CaP t}^{-1}$ ), which are the most  
19 valuable for plant nutrition. Enough evidence was obtained to claim that the ongoing technological  
20 race to meet the P discharge standards at the lowest cost possible should also reflect the agronomic  
21 value of P to plant nutrition to increase its competitiveness.

22

23 Keywords

24 wastewater management; circular economy; phosphorus; bioeconomy; techno-economic

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- 26 Nomenclature
- 27  $AHC_{10-40}$  = aromatic hydrocarbon compounds containing 10 - 40 hydrocarbon atoms
- 28 AIP = aluminum phosphates
- 29 ATA = acute toxicity to freshwater algae *Desmodesmus subspicatus*
- 30 ATD = acute toxicity to daphnia *Daphnia subspicatus*
- 31 ATF = acute toxicity to fish *Poecilla reticulata*
- 32 ATS = acute toxicity to seeds *Sinapis alba L.*
- 33 BET<sub>sa</sub> = Brunauere-Emmette-Teller surface area
- 34 BOD = biological oxygen demand
- 35 CaP = calcium phosphates (CaP1; CaP2 and CaP3)
- 36 CE = citizen equivalent
- 37 CEC = cation exchange capacity
- 38  $COD_{Cr}$  = chemical oxygen demand
- 39 DW = dry weight
- 40 EC = electric conductivity
- 41 EOHs = extractable organic halogens
- 42  $E_{sa}$  = external surface area
- 43 FeP = ferric phosphates
- 44 HV = heating value
- 45 LP = labile pool of carbon
- 46  $L_{sa}$  = Langmuir surface area
- 47  $MP_{sa}$  = micropore surface area
- 48  $P_{ext}$  = extractable P
- 49 RP = resistant pool of C

50 SPsa = single-point surface area

51 TSS = total suspended solids

52 ULD = under limit of detection

53

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54 Highlights

- 55 • Broken cellulose casings are readily available and undemanding to be charred because of its high  
56 heating value
- 57 • Activation of charred cellulose casings via calcium chloride is costless provided that the product  
58 is used for fertilization purposes anyway
- 59 • The sorbent obtained can capture more than 31 kg P t<sup>-1</sup> which is present in agronomically  
60 valuable forms (calcium phosphates)



61 • Introduction

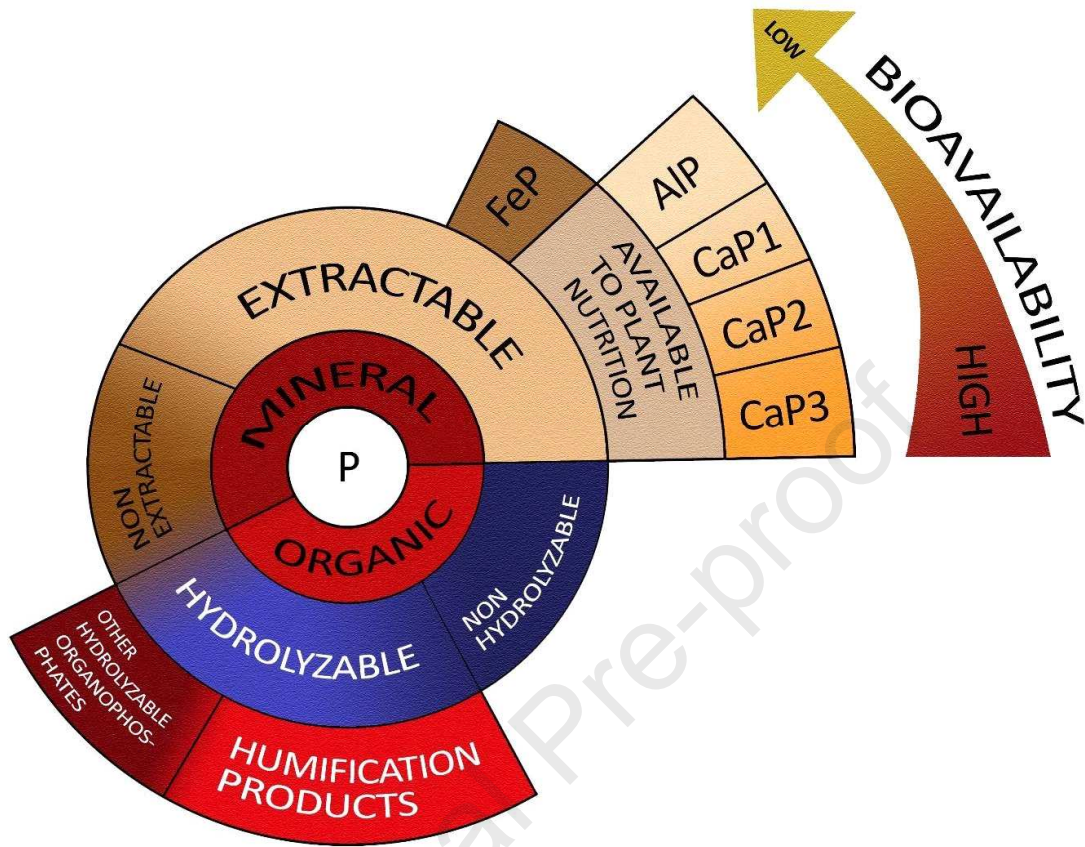
62 1.1 Scarcity of P available to plant nutrition

63 An adult person produces an average of 1.5 L urine day<sup>-1</sup>, which contains 3 - 5 g of phosphates  
64 (PO<sub>4</sub>). Proportionately, an average agglomeration with 1 million inhabitants produces some 1.5  
65 thousand m<sup>3</sup> of urine day<sup>-1</sup> with some 4 t of PO<sub>4</sub>(Dox et al., 2019). It should be noted that this urine  
66 is diluted in another 5 to 30 k m<sup>3</sup> wastewater according to local water consumption habits (Guan et  
67 al., 2020), which worsens things both technically and economically (Kliestik et al., 2020a).  
68 Worldwide wastewater production is growing rapidly, and excessive P discharge into water bodies  
69 intensifies eutrophication, leading to a wide range of environmental (Chen et al., 2020) and  
70 financial losses (Peters et al., 2020). Increasing P scarcity combined with its uneven geographical  
71 distribution is of growing concern since independent projections show economically viable P  
72 reserves becoming depleted within a few decades (Law and Pagilla, 2019). The total P content in  
73 arable soil ranges from 0.02 to 0.05 % (more in soils rich in organic matter), though ¾ of the  
74 world's soils have low P content (less than 0.01) and are therefore not suitable for plant production  
75 (Delgadillo-Velasco et al., 2020). Over 90 % of mined P is used for fertilization (Nättorp et al.,  
76 2017), whereas ¾ of P rocks are located in Morocco alone (de Boer et al. 2019). Only 1 P mine is  
77 currently operated in the EU, making its Member States 92% dependent on P import (over 1.2 M t  
78 year<sup>-1</sup>, Rowland et al., 2019), though purchasing power makes P affordability a limiting factor for  
79 agriculture in the developing world (Maroušek et al., 2020). Since there is a wide consensus that the  
80 economic importance of P will grow in the future (Stehel et al., 2019), a plethora of P recovery  
81 technologies (membrane bioreactors; deep-bed granular media filters; sorption; adsorption;  
82 coagulation; magnetically recoverable nanocomposites; ion-exchange resins, leaching of ashes

83 etc.) have been designed over previous decades (Peng et al., 2018a). In particular, membrane  
84 bioreactors were initially promising (Bai et al., 2016). However, membrane fouling appeared to be  
85 a critical obstacle to its widespread implementation as it decreases membrane lifetime and  
86 increases energy requirements for sludge water circulation (Bashar et al., 2018). Currently, the  
87 most widespread method for P recovery from sludge water is struvite ( $\text{MgNH}_4\text{PO}_4 \times 6\text{H}_2\text{O}$ )  
88 precipitation, which can capture up to 90 % of  $\text{PO}_4$  which refers to almost 85 % of total amount of  
89 P. Calculations by Nättorp et al. (2017) clearly demonstrated that the specific cost per P recovered  
90 (4 up to 10 €  $\text{kg}^{-1}$  P, without taking into account its suitability for fertilization) is not nowadays  
91 competitive with conventional fertilizer production (1.6 €  $\text{kg}^{-1}$  P), whereas the precipitation of  
92  $\text{PO}_4$  with  $\text{Fe}^{3+}$ ,  $\text{Al}^{3+}$ ,  $\text{Fe}^{2+}$  or  $\text{Ca}^{2+}$  salts into hydroxylapatites; struvite, or vivianite, are currently the  
93 least expensive (with struvite precipitation being the less expensive) and are therefore the most  
94 commonly used to comply with legal limits (Preisner et al., 2020). Acquisition costs of 1.3 - 2.5 M  
95 € 1 M citizen equivalent<sup>1</sup> (CE) and rising prices of ferric salts (chlorides, sulfates, or nitrates all  
96 above 1 k €  $\text{t}^{-1}$ ) are the main reasons why struvite production currently present cost rather than  
97 revenues. To make matters worse, the pH value has to be adjusted for struvite precipitation as well,  
98 which results in additional running costs. However, an underestimated economic and  
99 environmental threat is the low bioavailability of P that is captured in these complex minerals  
100 (Kiilerich et al., 2018). There are initial indications that prevailing precipitation techniques turn  
101  $\text{PO}_4$  into strong bonds with  $\text{Al}^{3+}$  and  $\text{Fe}^{3+}$  ions, which significantly reduces its agrochemical use  
102 (Tian et al., 2016). Suppliers consider this disadvantage to be a targeted feature, stating that  
103 long-lasting P fertilizers are the aim of the production. However, experienced farmers are no longer  
104 buying it. There is an increasing awareness that P in these minerals is mostly present in highly  
105 crystalline forms that need to first be broken down by soil microorganisms to become available to  
106 plants as nutrition. Low accessibility of struvite P to plant nutrition is associated with the

107 complexity of struvite nucleation and crystal morphology during crystal growth (driven by pH, ion  
108 speciation, temperature, and concentration of ions) which hinders its subsequent decomposition  
109 (Tansel et al., 2018). This process takes decades, and only under conditions under which the soil is  
110 optimally irrigated and rich in varied inputs of organic matter. Otherwise, the release of  
111 agronomically valuable forms of P takes hundreds of years (Ivanov et al., 2009). Multiple  
112 alternatives to improve the availability of nutrients locked in hydroxylapatites, struvite, vivianite,  
113 and other mineralization products can be traced in existing literature (Tansel et al., 2018).  
114 Nevertheless, all of them require costly chemicals or energy-intensive inputs (Chen et al., 2020).  
115 With regards to Fig. 1, it should be remarked that crops can only use calcium-based phosphates  
116 (CaP), and to a limited extent, also some aluminum based phosphates (AlP), whereas P present in  
117 ferric-based phosphates (FeP) and other forms can be considered inaccessible from the perspective  
118 of human life (Kulhánek et al., 2019).

119



120

121 Fig. 1: Classification of soil phosphorus according to the current state of knowledge (source:  
 122 authors).

123

## 124 1.2 Economics of bio-based sorbent production

125 Various types of biological waste have already been investigated to serve as P sorbents on a  
 126 commercial scale (De Gisi et al., 2016). Saha et al. (2019) reported that keratin and chitosan  
 127 showed some limited sorption properties on their own. Nevertheless, biomass is instead being  
 128 charred to increase its sorption properties nowadays, prolonging its durability, in particular, in  
 129 order to increase its economic attractiveness (90 - 230 € t<sup>-1</sup>) for subsequent use in conventional crop  
 130 production (Maroušek et al., 2019a). Farmers call these dusty carbonaceous products 'biochars'

131 and value them for reducing soil density, increasing water retention, and promoting nutrient  
132 management (Smetanová et al., 2013). Environmentalists point out that biochar application can be  
133 understood as one of the few profitable carbon sequestration methods, whereas the planting of trees  
134 is the most reasonable one (Luz et al., 2018). It is advantageous if the feedstock is biowaste for the  
135 disposal of which processing fees can be obtained (in the EU, the minimum fee for biowaste  
136 landfilling is now around 30 € t<sup>-1</sup>). It is further preferred if the feedstock's heating value is high  
137 enough so that it can pyrolyze itself (Lenhard, et al., 2019), or at least if the energy demands can be  
138 obtained from external waste energy source (combustion of methane from landfills, waste heat  
139 from industrial combustion engines etc.) to save on costs (Mardoyan and Braun, 2015). Provided  
140 that biochar is used as a soil enhancer, it is advantageous if any reactants used are also  
141 agronomically valuable so that its cost does not directly affect/increase the cost of P capture.

142

### 143 1.3 Proposal and research hypothesis

144 No profitable technology to facilitate the capture of P from sludge water is known to wastewater  
145 treatment plants (Peng et al., 2018a, b). As a result, developing countries have limited motivation  
146 for P capture and are interested only in unassuming methods like biosorbents that provide the  
147 benefits linked with P recovery for agriculture (Nättorp et al., 2017). Yang et al. (2018) used  
148 iron-modified chars made of from wastewater sludge. However, the energy requirements for  
149 sludge dewatering and pyrolysis are so high that they make the process economically unfeasible.  
150 Micháleková-Richveisová et al. (2017) proposed to hydrolyze biowaste by iron(III) nitrate  
151 (Fe(NO<sub>3</sub>)<sub>3</sub>). However, this activation proved too costly and dangerous for adaptation on  
152 commercial scale. Ajmal et al. (2020) reported an almost double increase in porosity after  
153 magnetization of biobased sorbents. However, similar concepts require robust modifications to  
154 current wastewater treatment technologies while the financial point-of-view remains unclear.

155 Given that the characteristics of commonly used biomass sorbents vary during the year, artificial  
156 casings made of regenerated cellulose are proposed as novel feedstock of constant quality. Since  
157 the demand for semi-finished meat products is growing globally, broken cellulose casings have  
158 become freely available in large quantities worldwide (Gabiatti et al., 2020a). Cellulose casings are  
159 supplied to meat businesses in the form of pleated gags so that they can be used in high-speed  
160 automatic filling lines, which are able to produce tens of tons of products per hour. However, once  
161 a small crack occurs, the entire roll of casings must be discarded, which represents a large amount  
162 of waste. Cellulose casings are made from viscose, which is obtained from plant fibers rich in  
163 cellulose. These artificial casings are highly elastic, so they are often reinforced with extra plant  
164 fibers to maintain the desired shape (Gabiatti et al., 2020b). A combination of hardly biodegradable  
165 regenerated cellulose and complex surface treatments (anti-condensation, against water absorption  
166 etc.) does not allow for their use as cattle feed (risk of cattle suffocation) nor for bio-refining  
167 (Lazaroiu et al., 2020).

168  
169 It has been hypothesized that charred cellulose casings that are subsequently activated by calcium  
170 chloride ( $\text{CaCl}_2$ ) could improve the economy of P recovery from wastewater sludge.

171  
172 • Materials and Methods  
173 2.1 Sorbent production  
174 Mechanically damaged casings from regenerated cellulose (see the characteristics in Table 1)  
175 were obtained from ViskoTeepak, s.r.o. (the Czech Republic) and stored in resealable cartons at 22  
176 °C with 60% relative humidity until their processing. This biowaste was fed into the UHL-07  
177 continuous pyrolysis apparatus (Aivotec, s.r.o., the Czech Republic) and charred at 320 °C (no  
178 additional energy source used, the process takes 6 minutes). The carbonaceous dust (see the

179 characteristics in Table 2) obtained was placed into the mixing reactor and mixed with 10% CaCl<sub>2</sub>  
180 (FICHEMA s.r.o., the Czech Republic) for 2 hours.

181

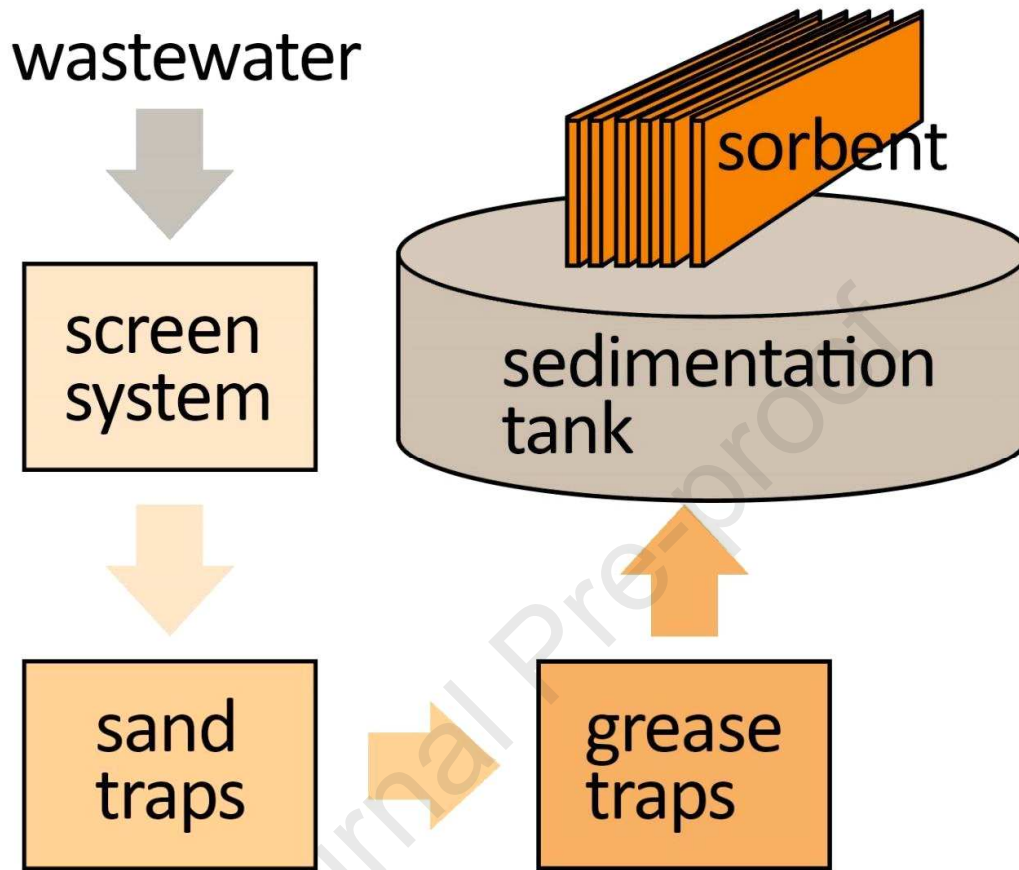
182 • Struvite production

183 Struvite precipitate was obtained by slowly mixing (20 hours at 5 rpm) magnesium chloride  
184 hexahydrate (MgCl<sub>2</sub> x 6 H<sub>2</sub>O, PENTA s.r.o., the Czech Republic) with oxygenated sludge water  
185 (200 L O<sub>2</sub> minute<sup>-1</sup> m<sup>-3</sup>) to obtain Mg : P ratio of 1.3 : 1 (molar weight). Subsequently, the ratio of N  
186 : P was adjusted to 3 : 1 (molar weight) by ammonia (NH<sub>3</sub>, PENTA s.r.o, the Czech Republic).  
187 Precipitate was separated by centrifugation (450 rpm for 1 hour) that was followed by 24 h of  
188 drying at 40 °C.

189

190 • Sorption experiment

191 Sorption experiments took place in the sedimentation tank of Strakonice wastewater treatment  
192 plant (75 k CE ~ 3.0 M m<sup>3</sup>y<sup>-1</sup>, the Czech Republic) according to the schema showed in Fig. 2.  
193 Sorbent was filled into bales (3 m \* 0.1 m \* 1 m) made of permeable fabric to ensure a good  
194 reaction surface. The bales filled with sorbent were immersed into the sedimentation tank<sup>-1</sup>).



195

196 Fig. 2: Setup of the wastewater treatment plant where the experiment took place.

197

198 • Analytical procedures

199 Dry weight (DW) was analyzed after 3 days at 105 °C in the Goldbrunn1450 vacuum (140 Pa)

200 dryer (Goldbrunn therm, Germany) by the EP 320M weights (Precisa Gravimetrics AG,

201 Switzerland). The thickness of the casings walls was measured using the SBS-CT-123 PRO



202 coating thickness gauge meter (Steinberg systems, Germany). The cellulose content was analyzed  
203 using the FIBRE THERM FibreBag technology (C. Gerhardt GmbH & Co. KG, Germany). Oils  
204 and fats were quantified using the Soxhlet extractor (Corning Inc., NY, USA) and analytical  
205 hexane (99.9 %). Labile (LP) and resistant (RP) pools of carbon (resistance to acid hydrolysis)  
206 were analyzed with sulfur acid ( $H_2SO_4$ ) according to Shirato and Yokozawa (2006), using the  
207 EA5100 elemental analysis (Analytik Jena AG, Germany). The heating value (HV) was analyzed  
208 using the CAL3K-A bomb calorimeter system (Digital Data Systems Ltd., South Africa). Ash  
209 content was determined according to D1762-84 (ASTM). The Seven2GoPro multimeter  
210 (Mettler-Toledo, Switzerland) was used to analyze pH and electrical conductivity (EC). The  
211 cation-exchange-capacity (CEC) was analyzed according to the protocol proposed by  
212 Munera-Echeverri et al. (2018). Analyses for lead (Pb), cadmium (Cd), copper (Cu), nickel (Ni),  
213 mercury (Hg), zinc (Zn), chromium (Cr), boron (B), manganese (Mn), sulfur (S), naphthalene,  
214 acenaphthylene, acenaphthene, fluorene, phenanthrene, anthracene, fluoranthrene, pyrene,  
215 benzo(a)anthracene, chrysene, benzo(b)fluoranthrene, benzo(k)fluoranthrene, benzo(a)pyrene,  
216 indeno(1,2,3-cd)pyrene, dibenzo(a,h)anthracene, benzo(g,h,i)perylene, benzene, toluene,  
217 ethylbenzene, m- xylene, p- xylene, o-xylene, naphthalene, extractable organic halogens (EOHs),  
218 and aromatic hydrocarbon compounds containing 10 - 40 hydrocarbon atoms ( $AHC_{10-40}$ ) were  
219 carried out as described in Vochozka et al. (2016a). Acute toxicity to fish *Poecilla reticulata*  
220 (ATF); daphnia *Daphnia subspicatus* (ATD), freshwater algae *Desmodesmus subspicatus* (ATA),  
221 and seeds *Sinapis alba L.* (ATS) was analyzed according to Vochozka et al. (2016b). The  
222 Brunauer-Emmett-Teller surface area (BET<sub>sa</sub>); single-point surface area (partial vapor pressure  
223 of helium in equilibrium with the surface)/(saturated pressure of helium) = 0.1 (SP<sub>sa</sub>); Langmuir  
224 surface area (L<sub>sa</sub>); micropore surface area (MP<sub>sa</sub>) and external surface areas (E<sub>sa</sub>) were analyzed  
225 by the 3H-2000PS4 porosimeter (Tmax BE, China) after 48 hours of degassing at 300 °C.

226 Extractable P ( $P_{\text{ext}}$ ) and its subsequent fractionation (FeP, AlP, CaP1, CaP2, and CaP3), biological  
227 oxygen demand (BOD), chemical oxygen demand ( $\text{COD}_{\text{Cr}}$ ), and total suspended solids (TSS) were  
228 analyzed as described in Maroušek et al. (2019b). Cost breakdown was carried out as proposed by  
229 Hašková (2017), and all the costs are rounded up.

230

### 231 3. Results and Discussion

#### 232 3.1 Waste cellulose casings characteristics

233 Analyses on waste cellulose casings (Table 1) showed high proportion of cellulose fiber and high  
234 proportion of RP which can be interpreted as a high resistance to  $\text{H}_2\text{SO}_4$  hydrolysis (Shirato and  
235 Yokozawa, 2006). These findings are in accordance with Gabiatti et al. (2020a), who adds that the  
236 combination of highly crystalline cellulose and complex surface treatments (esters of glycerol,  
237 polyglycerol, sorbitan esters and alcohol ethoxylates are sprayed on the casing's surface or mixed  
238 into the matrix of regenerated cellulose) are used to make the fibrous casings resistant microbial  
239 attacks. Such a design makes sugar-based biorefining options economically problematic (Gabiatti  
240 et al. 2020b). The combination of cellulose, energy-rich oils, and fats makes the HV slightly higher  
241 than other cellulose-based feedstock like paper or cardboard (Vochozka et al., 2016c). Provided  
242 that pyrolysis is a process of exothermic nature, this allowed to achieve self-sufficient pyrolysis at  
243 temperature of 320 °C. Ronsse et al. (2013) reports that such low temperatures are limiting to the  
244 formation of good sorption properties. However, an external energy source is not a viable  
245 alternative for economic reasons (Jandačka et al., 2017). Provided that cellulose casings made by  
246 other manufacturers are made from similar raw materials (Kliestik et al., 2020b), of similar  
247 thickness, and via similar procedures (Peters et al., 2020), it is not surprising that they show similar  
248 properties (Maroušek et al., 2015). It can be argued that these waste cellulose casings comply with

249 globally established commercial standards and allow for valid generalizations of subsequently  
 250 acquired knowledge.

251

252 Table 1: Characteristics of the cellulose casings (n = 12;  $\alpha = 0.1$ ).

253

Compounds/Property	Characteristics (unit)
Cellulose	$809.3 \pm 15.6$ (g kg <sup>-1</sup> )
Oils and fats	$155.1 \pm 10.2$ (g kg <sup>-1</sup> )
HV	$16.9 \pm 0.3$ (MJ kg <sup>-1</sup> )
Casing wall thickness	$92.3 \pm 44.5$ ( $\mu$ m)
LP	2 (% cellulose)
RP	98 (% cellulose)

254

255

### 256 3.2 Sorbent characteristics

257 In regards to charred cellulose casings, the ash content (Table 2) is as low as it is in chars made of  
 258 hardly biodegradable phytomass like virgin paper (Vochozka et al., 2016c) or fermentation  
 259 residues (Mardoyan and Braun, 2015), which consist mainly of cellulose with a high degree of  
 260 polymerization. Low ash content also indicates that the pyrolytic process did not suffer from  
 261 defects, since it is probable that only oxygen (O<sub>2</sub>) from the air capsules present in the feedstock  
 262 entered the pyrolysis chamber. The pH level of the char obtained is average compared to pyrolyzed  
 263 terrestrial phytomass (Ronsse et al., 2013), and this fact confirms that the cellulose casings were  
 264 not made of terrestrial algae or seaweed. EC is about 50 % lower than found in commercially  
 265 available chars as reviewed by Saha et al. (2019). Low EC indicates that salt crystals remained

266 locked in char aggregates, which is a highly likely to be a consequence of low temperature during  
267 pyrolysis (Smetanová et al., 2013). From this, it can be implicitly assumed that it may be useful in  
268 the future to investigate whether it would be beneficial to modify this sorbent mechanically, for  
269 example, by crushing (Hadzima et al., 2007). One of the latest protocols for biochar CEC analysis  
270 (Munera-Echeverri et al., 2018) was applied and looks to be favorably reproducible (as this used to  
271 be a problem with earlier methods). However, given that the method by Munera-Echeverri et al.  
272 (2018) is quite new, the results reported here are not comparable with prior papers. The levels of  
273 extractable Ca, Mg and K are average in comparison with charred sorbents of biowaste origin as  
274 reviewed by Li et al. (2017). Data in Table 2 also shows that all crucial heavy metals and other  
275 critical substances present are either under limit of detection (ULD) or well below soil hygiene  
276 regulations (De Gisi et al., 2016), which reflects the fact that cellulose casings are intended to be in  
277 contact with food and must therefore comply with corresponding laws (Gabiatti et al., 2020a).  
278 When compared to Luz et al. (2018), only low levels of polycyclic aromatic hydrocarbons could be  
279 traced, which was indirectly confirmed by no signs of acute biotoxicity. It follows from the above  
280 that charred casings from waste cellulose casings can be used without any obstacles as a soil  
281 enhancer. In regard to microporosity, the results are about two-thirds lower than reported by Li et  
282 al. (2017). Nevertheless, even a twofold increase in porosity does not guarantee improved results in  
283 P capture (Ajmal et al., 2020). In agreement with Ronsse et al. (2013) one can guess that this is due  
284 to the low temperatures of the pyrolysis process (Kovacova et al., 2019) which are caused by the  
285 effort for minimal cost (no external energy source). Most importantly, CaP is some 82 % of  $P_{\text{ext}}$ . As  
286 far as the sorbent production is concerned, it should be remembered that the technology includes  
287 not only the UHL-07 pyrolysis unit but also the tailor-made mixing reactor and a dipping system.  
288 The guaranteed lifespan of all devices is 12 years, and the total price is some 0.8 M €. The  
289 technology enables the production of 100 kg char hour<sup>-1</sup> for 8.7 k hours year<sup>-1</sup> and the overhead is

290 about 5 k € year<sup>-1</sup>. Although the heating value of the feedstock is sufficient for self-pyrolysis, the  
 291 pyrolysis unit uses natural gas at the very start of the process and to occasionally stabilize the  
 292 quality of the pyrolytic gas. The cost of natural gas is 1 € char t<sup>-1</sup> on average. The mixing reactor  
 293 that activates the sorbent with 10 % CaCl<sub>2</sub> needs an electrical supply of some 1.5 € char t<sup>-1</sup>, the  
 294 immersion system and other handling requires energy of 0.5 € char t<sup>-1</sup>. 1 t of char is activated by  
 295 160 kg of 10 % CaCl<sub>2</sub> (this number seems to be an operationally achievable state of equilibrium;  
 296 more CaCl<sub>2</sub> has never been captured), whereas the wholesale price of 77 % CaCl<sub>2</sub> is about 90 € t<sup>-1</sup>.  
 297

298 Table 2: Characteristics of charred cellulose casings (n = 10;  $\alpha = 0.1$ ) indicate that low pyrolysis  
 299 temperature might be limiting for its sorption properties (ULD = under limit of detection, (N/A =  
 300 not applicable).

301

Compounds/Property	Sorbent	Sludge water	Struvite
Density (g L <sup>-1</sup> )	631 ± 15.8	1072.6 ± 22.9	1690.3 ± 16.1
Ash (g kg <sup>-1</sup> )	7.8 ± 4.2	N/A	N/A
BOD (mg L <sup>-1</sup> )	N/A	1093.7 ± 64.8	N/A
COD <sub>Cr</sub> (mg L <sup>-1</sup> )	N/A	2506.4 ± 113.9	N/A
pH	8.1 ± 1.0	8.4 ± 0.5	7.9 ± 0.3
EC (mS cm <sup>-1</sup> )	33.4 ± 7.5	1.8 ± 0.4	N/A
CEC (mol H <sup>+</sup> kg <sup>-1</sup> )	50.7 ± 3.8	N/A	N/A
Pb (ppm)	ULD	ULD	0.2 ± 0.1
Cd (ppm)	ULD	ULD	0.2 ± 0.1
Cu (ppm)	0.2 ± 0.1	0.1 ± 0.1	0.3 ± 0.1

Ni (ppm)	0.3 ± 0.1	ULD	ULD
Hg (ppm)	ULD	ULD	ULD
Zn (ppm)	4.5 ± 2.2	0.1 ± 2.2	25.7 ± 3.5
Cr (ppm)	ULD	ULD	0.5 ± 0.2
B (ppm)	ULD	ULD	ULD
Mn (ppm)	17.2 ± 3.9	ULD	33.6 ± 15.7
S (ppm)	21.5 ± 6.6	ULD	ULD
Naphthalene (ppm)	ULD	ULD	ULD
Acenaphthylene (ppm)	ULD	ULD	ULD
Acenaphthene (ppm)	ULD	ULD	ULD
Fluorene (ppm)	ULD	ULD	ULD
Phenanthrene (ppm)	ULD	ULD	ULD
Anthracene (ppm)	ULD	ULD	ULD
Fluoranthrene (ppm)	ULD	ULD	ULD
Pyrene (ppm)	1.1 ± 0.4	ULD	ULD
Benzo(a)anthracene (ppm)	0.3 ± 0.3	ULD	ULD
Chrysene (ppm)	ULD	ULD	ULD
Benzo(b)fluoranthrene (ppm)	ULD	ULD	ULD
Benzo(a)pyrene (ppm)	2.6 ± 1.7	ULD	ULD
Indeno(1,2,3-cd)pyrene (ppm)	ULD	ULD	ULD
Dibenzo(a,h)anthracene (ppm)	ULD	ULD	ULD
Benzo(g,h,i)perylene (ppm)	ULD	ULD	ULD
Benzene (ppm)	0.7 ± 0.5	ULD	ULD

Toluene (ppm)	ULD	ULD	ULD
Ethylbenzene (ppm)	ULD	ULD	ULD
M-xylene (ppm)	ULD	ULD	ULD
P-xylene (ppm)	ULD	ULD	ULD
O-xylene (ppm)	ULD	ULD	ULD
Naphthalene (ppm)	ULD	ULD	ULD
EOHs (ppm)	ULD	ULD	ULD
AHC <sub>10-40</sub> (ppm)	ULD	ULD	ULD
ATF (%)	ULD	10	N/A
ATD (%)	ULD	5	N/A
ATA (%)	ULD	ULD	N/A
ATS (%)	ULD	ULD	N/A
BET <sub>sa</sub> (m <sup>2</sup> g <sup>-1</sup> )	172.8 ± 93.5	N/A	N/A
L <sub>sa</sub> (m <sup>2</sup> g <sup>-1</sup> )	25.7 ± 14.6	N/A	N/A
SP <sub>sa</sub> (m <sup>2</sup> g <sup>-1</sup> )	19.0 ± 13.9	N/A	N/A
MP <sub>sa</sub> (m <sup>2</sup> g <sup>-1</sup> )	19.8 ± 15.9	N/A	N/A
E <sub>sa</sub> (m <sup>2</sup> g <sup>-1</sup> )	16.1 ± 11.3	N/A	N/A
P <sub>ext</sub> (mg kg <sup>-1</sup> )	233.5 ± 40.3	N/A	250.3 ± 31.8
P <sub>ext</sub> (mg L <sup>-1</sup> )	N/A	52.5 ± 6.0	N/A
FeP (% P <sub>ext</sub> )	8	5	25
AIP (% P <sub>ext</sub> )	10	16	36
CaP1 (% P <sub>ext</sub> )	31	13	17
CaP2 (% P <sub>ext</sub> )	22	25	13

CaP3 (% $P_{\text{ext}}$ )	29	41	9
Depreciation and overheads (€ $t^{-1}$ )	83	N/A	66
Energy cost (€ $t^{-1}$ )	3	N/A	13
Reactants (€ $t^{-1}$ )	2	N/A	25

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302

303

## 304 3.3 Sludge water characteristics

305 In terms of sludge water, its density and pH (Table 2) is similar to what was reported by Tansel et  
306 al. (2018) and other studies (Preisner et al., 2020; Saha et al., 2019) around the world. BOD,  
307  $COD_{Cr}$ , DM, TSS, and EC are also in the usual range (Peng et al., 2018a, b). Crucial heavy metals  
308 and other critical pollutants were at almost negligible levels (see Chen et al., 2020) and only mild  
309 acute biotoxicity was detected (ATF and ATD). With regards to Vochozka et al. (2016b), it can be  
310 argued that the data obtained on acute biotoxicity is a consequence of the combination of BOD and  
311 microbial load. Nevertheless, such results do not deviate in any way from the common sludge  
312 water values. The level of P is about one seventh higher in comparison with other similarly  
313 populated locations (Chen et al., 2020), and it can be concluded that such a high level is probably  
314 not solely of urinary origin (Dox et al., 2019). Following that the less accessible forms of P (in  
315 particular FeP and AIP) are also represented, it can be assumed that the above-average values of P  
316 might originate from local industry or an increased use of detergents containing P (Kiilerich et al.,  
317 2018), which may be associated with the ongoing pandemic. Following the above, this sludge  
318 water is a sample that shows characteristics similar rather to large agglomerations with higher level  
319 of industry. The values of P also prove that sludge water did not undergo any P decrease procedure  
320 which are usually required before being discharged into water systems (Guan et al., 2020). It is  
321 appropriate to note that only 21 % of  $P_{\text{ext}}$  is present in forms that are hardly available to plant



322 nutrition. All in all, it can be stated that these characteristics of sludge water are in agreement with  
323 other literature (Chen et al., 2020) and the lessons learned are potentially widely transferable. The  
324 investigated sludge water is by no means being modified; therefore its pricing is slightly negative  
325 since getting rid of it means a cost savings for wastewater plants (Kiilerich et al., 2018). However,  
326 in many countries, the application of sludge water to soil is subject to various restrictions (it cannot  
327 be performed if the fields are under snow, if the soil is frozen, in case of sandy soil, if there are  
328 water systems nearby etc.) and in some countries, it is even prohibited (Bashar et al., 2018).

329

### 330 3.4 Struvite characteristics

331 In terms of the struvite produced, the pH obtained is similar to commercially available products,  
332 such as those reviewed by Tansel et al. (2018); therefore, soil acidification should not be expected.  
333 The concentration of life-threatening heavy metals and other hazardous compounds is generally  
334 higher than that of the activated sorbent or wastewater, which corresponds to the method of  
335 production and an increased concentration of P. Nevertheless, these figures are similar to those  
336 found in commercially available products and meet all the regulations necessary to be used as  
337 fertilizer (Nättorp et al., 2017). From a biological point of view, it is important to note that only  
338 some 39 % of  $P_{\text{ext}}$  is present in CaP forms. A struvite reactor, centrifuge, and dryer are mandatory  
339 (in total 0.5 M € with guaranteed service life of 6years) to run struvite production 8.7 k hours  
340  $\text{year}^{-1}$ . The overhead is some 80 k €  $\text{year}^{-1}$ . The energy costs of operating the reactor is  
341 approximately 4 € struvite  $\text{t}^{-1}$ , centrifugation 3 € struvite  $\text{t}^{-1}$  and drying 6 € struvite  $\text{t}^{-1}$ .  $\text{MgCl}_2 \times 6$   
342  $\text{H}_2\text{O}$  (21 €  $\text{t}^{-1}$ ),  $\text{NH}_4$  (3 €  $\text{t}^{-1}$ ) and  $\text{O}_2$  (1 €  $\text{kg}^{-1}$ ) are needed to run the process.

343

344

### 345 3.5 Economic insights

346 Wastewater treatment plants pressingly call for a P capture process that is undemanding in terms of  
347 acquisition and running costs and is capable of producing competitive fertilizers which meet all the  
348 legal requirements (Peng et al., 2018a). Bashar et al. (2018) identified that the Integrated  
349 Fixed-Film Activated Sludge Systems with Enhanced Biological Phosphorus removal (also known  
350 as IFAS-EBPR) is the most cost-effective system for P removal in the USA ( $60 \text{ € P kg}^{-1}$ ), whereas  
351 precipitation of struvite ( $52 \text{ € P kg}^{-1}$ ) seems to be of the lowest cost in the EU. The economics of  
352 struvite production in this study is similar to other commercial production, provided that the  
353 production costs calculated (Table 3) during the pilot testing are only slightly lower than the lowest  
354 struvite wholesale prices (Tansel et al., 2018). Low margins are indirectly signaling that there is a  
355 little demand for this product. Yang et al. (2018) used iron-activated charred sludge as a sorbent.  
356 However, this concept proved unfeasible due to high energy demands since 1/ huge amounts of  
357 water needs to be removed from the sludge before it can be pyrolyzed; and 2/ a lot of external  
358 energy is needed to pyrolyze sludge. Efforts to activate biowaste sorbents via iron(III) nitrate  
359 showed economically unrealistic and dangerous (Micháleková-Richveisová et al., 2017). A closer  
360 look at the cost analysis (Table 3) shows that even though struvite production is currently one of the  
361 most efficient P capture technologies worldwide, the novel sorbent can be competitive in some  
362 aspects. Discussion over the production costs of the sorbent opens wide economic polemics. Cost  
363 breakdown indicates that its production cost is somewhere in the neighborhood of  $90 \text{ € t}^{-1}$ , which is  
364 close to the cheapest charcoal produced from fermentation residues (Mardoyan and Braun, 2015).  
365 It depends on who (and how) runs the business. Theoretically, the business concept could fulfill all  
366 the theoretical concepts of circular economics. Provided that farmers buy biochar, P, and  $\text{CaCl}_2$   
367 anyway, one can argue that this is a win/win/win/win business where: A/ producers of meat  
368 products can get rid of their voluminous waste at lower costs; B/ wastewater treatment plants can  
369 capture P at lower costs; C/ farmers can get P at lower costs, and D/ humanity benefits from 1/

370 lower wastewater treatment cost; 2/ lower food production cost; 3/ carbon sequestration via  
 371 biochar; 4/ reduction of landfilling and prevention of greenhouse gas emissions; and 5/ increase in  
 372 food security due to the reduction of P imports and an increase in the quality of soil and water.  
 373 Nättorp et al. (2017) demonstrated that P obtained from P rocks is 1.6 € kg<sup>-1</sup> P. Data in Table 3  
 374 indicates that struvite production cost remains at just more than a tenth above this number and is  
 375 still not competitive with such prices. Nevertheless, the novel sorbent technology may seem  
 376 attractive, but it is more important to note that similar calculations can be misleading as they do not  
 377 reflect the biological nature of things (Kulhánek et al., 2019). Analyses on CaP indicate that  
 378 sorbents are capable of capturing these agronomically valuable P fractions with more than double  
 379 cost efficiency. Nevertheless, the desire to obtain an agronomically and biologically efficient  
 380 fertilizer is not a sufficient reason to make such a managerial decision because the cost of the  
 381 application must also be considered (Valaskova et al., 2020). A calculation-simulating application  
 382 of 1 kg CaP ha<sup>-1</sup> at 20 km distance confirms that the application of sludge water is irrational  
 383 because its P concentration is extremely low and therefore expensive (Skapa, 2012), not taking into  
 384 account possible legislative restrictions.

385  
 386 Table 3: Cost analysis indicates that conventional methods of capturing P from sludge water are  
 387 still not competitive with P fertilizers from P rocks. But things can change if the sorbent is  
 388 produced from biowaste and activated by agrochemicals.

389

Compounds/Property	Sorbent	Sludge water	Struvite
P (kg t <sup>-1</sup> )	31.8 ± 2.2	0.4 ± 0.1	57.0 ± 4.5
CaP (mg kg <sup>-1</sup> )	191	369	98

Production cost (€ t <sup>1</sup> )	88	0	104
Production cost (€ P kg <sup>1</sup> )	0.4	N/A	1.8
Production cost (€ CaP kg <sup>1</sup> )	461	N/A	1061
Application cost (€ kg CaP hā <sup>1</sup> )	128	9 255	305

390

## 391 4. Conclusion

392 Ignoring economic and legal realities, it can be presumed that the best source of P for plant  
 393 nutrition is sludge water. Despite this fact, the use of sludge water for plant nutrition is constrained  
 394 by costly logistics and is often subject to legal regulations. Struvite production is a proven  
 395 technology, and the cost of such captured P is now closer than ever to meeting the pricing of P from  
 396 conventional P fertilizers that are made of P rocks. However, one disadvantages of struvite is that  
 397 less than 40 % of its P<sub>ext</sub> is present in CaP forms which are the most valuable for plant nutrition. The  
 398 rest of the P<sub>ext</sub> in the struvite can become agronomically valuable only over decades, and only if the  
 399 optimal conditions for high activity of soil microorganisms are created in the soil.

400 A novel method of sorbent production from waste cellulose casings consisting of pyrolysis and  
 401 CaCl<sub>2</sub> activation was proposed and techno-economically investigated in a pilot scale. It was first  
 402 reported that this sorbent allows for the capture of 31.8 kg P t<sup>-1</sup>, whereas more than 80 % of the  
 403 captured P is present in forms that are easily available as crop nutrition. Due to the fact that  
 404 numerous cost saving measures and synergies were combined in the production of the sorbent, it  
 405 can be concluded that the novel sorbent could be financially promising in commercial scales.

406 Liquid biowaste from agriculture and food production is known for high P levels therefore, further

407 research could take this direction.

408

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412

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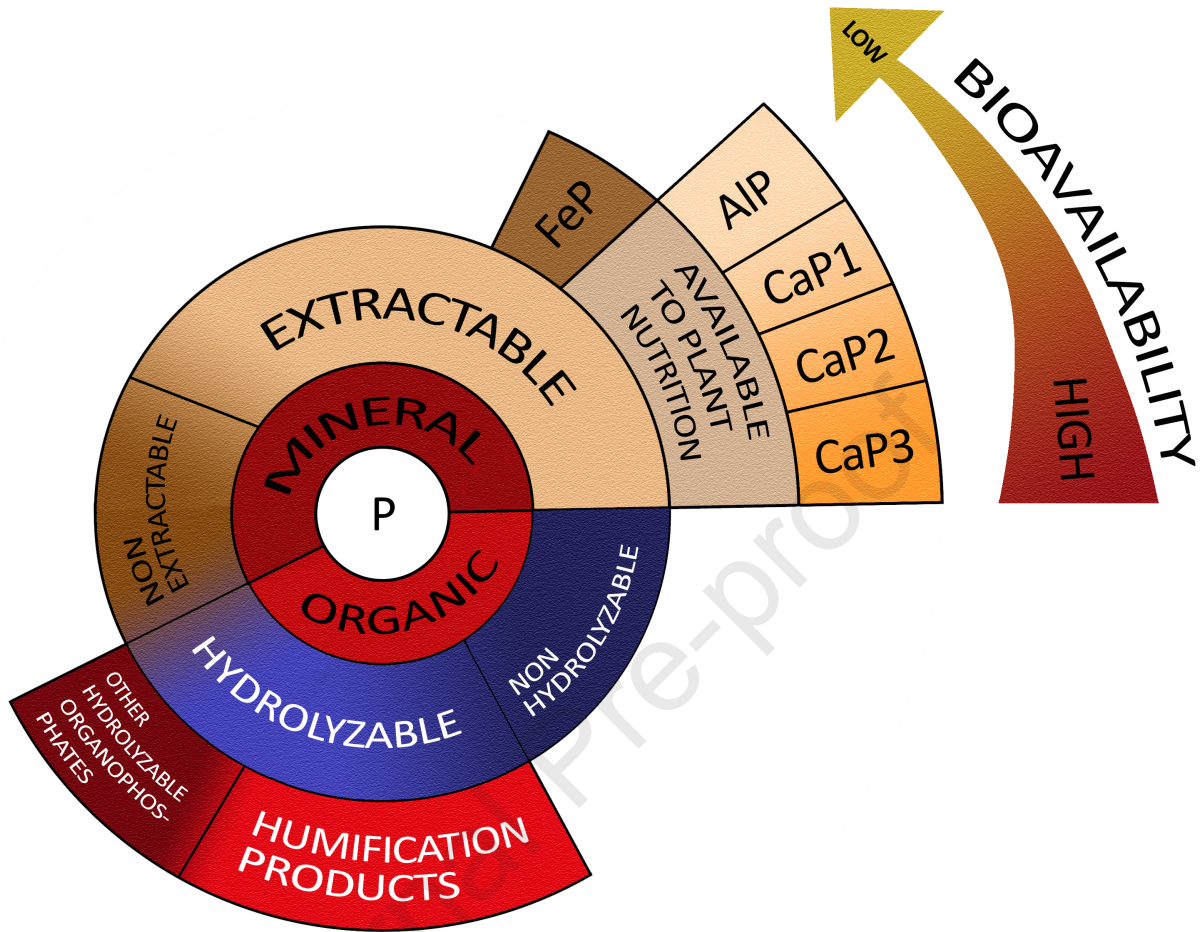
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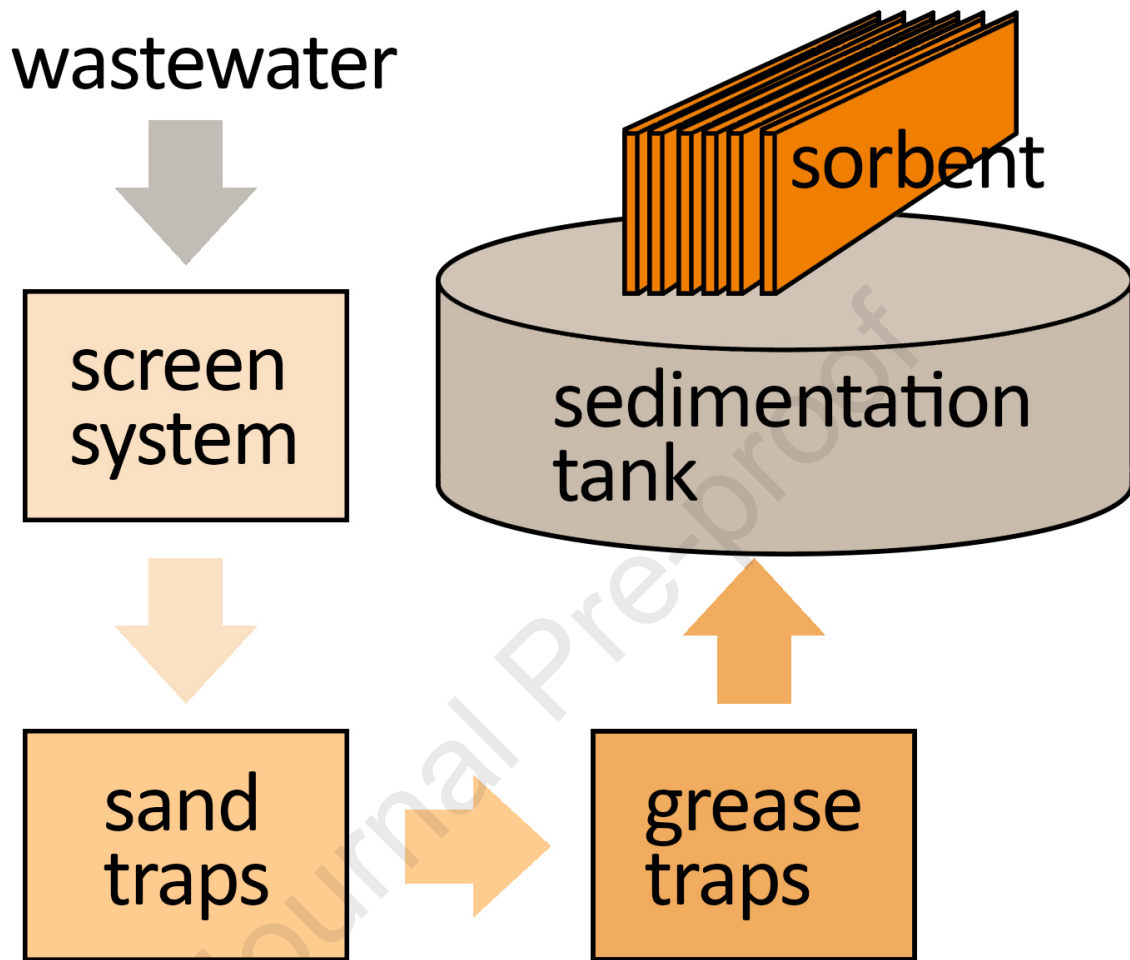
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## Highlights

- Charred and calcium chloride-activated broken cellulose casings are used for phosphates sorption
- Technology cost is low since it does not require external energy and reactant cost is irrelevant
- Bioavailability of captured phosphorus is high in comparison with struvite

**Declaration of Interest Statement**

Authors declare no conflict of interest.

Journal Pre-proof